

**PCT**WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 6 : <b>A01N 37/16, A01G 31/00 // (A01N 37/16, 37:16, 59:00)</b>		A1	(11) International Publication Number: <b>WO 99/51095</b> (43) International Publication Date: <b>14 October 1999 (14.10.99)</b>
(21) International Application Number: <b>PCT/US99/06010</b>		(81) Designated States: AE, AL, AM, AT, AT (Utility model), AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, CZ (Utility model), DE, DE (Utility model), DK, DK (Utility model), EE, EE (Utility model), ES, FI, FI (Utility model), GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SK (Utility model), SL, TJ, TM, TR, TT, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).	
(22) International Filing Date: <b>19 March 1999 (19.03.99)</b>		Published <i>With international search report.</i> <i>With amended claims.</i>	
(30) Priority Data: <b>09/055,609 6 April 1998 (06.04.98) US</b>			
(71) Applicant: ECOLAB INC. [US/US]; Ecolab Center, St. Paul, MN 55102 (US).			
(72) Inventors: HEI, Robert, D., P.; 6185 26th Street North, Oakdale, MN 55128 (US). ADKINS, Leanne, J.; 1220 Mourning Dove Court, Eagan, MN 55123 (US). LOKKESMOE, Keith, D.; 14463 Monterey Avenue, Savage, MN 55378 (US). HANSON, Heidi, M.; 3035 S. 34th Avenue, Minneapolis, MN 55406 (US). CORDS, Bruce, R.; 4635 Park Ridge Drive, Eagan, MN 55123 (US).			
(74) Agent: DAIGNAULT, Ronald, A.; Merchant, Gould, Smith, Edell, Welter & Schmidt, P.A., 3100 Norwest Center, 90 South Seventh Street, Minneapolis, MN 55402-4131 (US).			

(54) Title: PEROXY ACID TREATMENT TO CONTROL PATHOGENIC ORGANISMS ON GROWING PLANTS

## (57) Abstract

Both a process and a method of using peracid compositions, especially mixed peracid systems, to treat field or greenhouse grown plant tissue, seeds, fruits, and growing media and containers. The peracid can lower the natural, plant pathogen and human pathogenic microbial load resulting in less waste to molding, spoilage, and destruction because of pathogenic poisons.

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakhstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

**PEROXY ACID TREATMENT TO CONTROL  
PATHOGENIC ORGANISMS ON GROWING PLANTS**

**Field of the Invention**

5        The invention relates to a process of using peracid compositions, especially mixed peracid systems, to treat field, hydroponic or greenhouse growing plant tissue, seeds, fruits, growing media and containers. The peracid can lower the natural, plant pathogen and human pathogenic microbial load resulting in less waste to molding, spoilage, and destruction because of pathogenic poisons.

10

**Background of the Invention**

15        In the production of fruits and vegetables, plants can be grown in the field, in greenhouses, and hydroponically. Each location has its own growing medium, environment and growing conditions. Agricultural personnel work to maximize production by maximizing growing conditions while minimizing attack on seeds, seedlings, plants and fruit by living pests. Such pests include insects, rodents, bacteria, fungi, etc.

20        Substantial attention has been given to antimicrobial compounds that attack bacteria and fungi on seeds, seedlings, growing plants and fruit in the production cycle on growing plants. The use of fungicides in agriculture is necessitated by the great losses caused by a wide variety of plant-pathogenic microorganisms. To be economic, the costs of controlling plant diseases by the application of bactericides and fungicides must be offset by potential gains of several fold. Large tonages of fungicides are required in the agriculture of apples, pears, bananas, cereals, cocoa, 25 coffee, cotton, potatoes, tobacco, grapes, sprouts and other common fruits and vegetables including celery, leeks, onions, lettuce, spinach, brussel sprouts, potatoes, truffles, garlic, shallots, peppers, beans, tomatoes, almonds, pears, apples, peanuts and others. Fungicides are typically applied in water suspension with hydraulic sprayers or in the form of dust, granules or fumigants. Early fungicides 30 included sulfur and polysulfides, heavy metals and others. Such harsh fungicides have been replaced by newer but still toxic materials such as quinones, organosulfur compounds, imidazolines and guanidines, trichloromethylthiocarboximides, chlorinated and nitrated benzenes, oxithines, benzimidazoles, pyrimidines, and others. These broad spectrum protectant materials effect enzyme and membrane 35 systems of the target microorganism. Typically, the mode of action includes

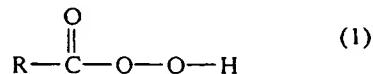
inhibition of fungal or bacterial energy production, interference with biosynthesis or disruption of cell membrane structure.

The above fungicides have had some success; however, they are viewed as toxic materials and a substantial quantity of plant produce is wasted due to their 5 deleterious effect. Accordingly, a substantial need exists to continue to develop antimicrobial materials that can protect growing plants including seeds, cuttings, seedlings, growing plants, plant parts, fruit, and other agricultural produce.

### Peroxy Acids

10 Further, human and plant pathogenic bacteria and fungi can be a contamination problem in growing plants. We have found coli form, salmonella, and other bacteria common in the agricultural and greenhouse environment can contaminate growing plants and pose a threat to human health in consumption of fresh vegetables, fruit and produce. A substantial need exists for treatments that can 15 reduce bacterial contamination.

Peroxy acids are strong oxidizers and have the simple general structure given as formula (1), where R can be essentially any hydrocarbon group:



20

### Antimicrobial Treatments

Peroxy-containing compositions are known for use in the production of 25 microbial agents. One such composition is disclosed in Bowing et al., U.S. Pat. No. 4,051,059 containing peracetic acid, acetic acid or mixtures of peracetic and acetic acid, hydrogen peroxide, anionic surface active compounds such as sulfonates and sulfates, and water.

Peracetic acid has been shown to be a good biocide, but only at fairly high concentrations (generally greater than 100 parts per million (ppm)). Similarly, peroxyfatty acids have also been shown to be biocidal, but only at high 30 concentrations (greater than 200 ppm), such as in the composition disclosed in European Patent Application No. 233,731.

GB 2187958 A and EU 0242990 A2 describe the use of either peracetic or per-propionic acid for controlling plant pathogens on flowers and fruit tissue. They are directed to edible field-grown plants and cereal crops.

WO 94/06294 describes the use of a single peracid composition along with 5 mixtures of aliphatic acids for vegetable disinfection.

U.S. 5,168,655 relates to hydroponic treatment using peracids. The reference describes peracid treatments of hydroponic growing substrates (e.g., rock wool) prior to growth; i.e., the growth substrate is treated after a crop production cycle and prior to a subsequent crop-production cycle. In contrast, the present invention describes 10 hydroponic treatment during the growth cycle.

U.S. Patent No. 5,200,189 to Oakes et al. describes use of mixed peracid compositions to enhance microbial kill for hard surface sanitizing. Certain mixed peracids have now been found useful to enhance microbial kill on sensitive growing plant tissue or its harvested fruiting matter.

15 U.S. Patent No. 2,512,640 to Greenspan et al. discloses the use of single peracid compositions to enhance microbial reduction on produce, reduce produce browning, and prevent spoilage. Greenspan does not disclose any mixed peracid synergies and applies the peracid only to harvested fruit.

GB 2257630A describes the use of a single peracid which is activated by an 20 activator (Fe, Cu, Br, I) for controlling microbial counts on hard surfaces, effluent waters and growing plant tissues. Again, this is a single peracid composition which fails to teach synergies between mixed peracids.

DK 9300538 describes the use of peracetic acid, followed by a biological combat, for controlling pathogens in recirculating watering systems to plant crops.

25 This reference does not describe any direct crop treatments.

JP 07031210 teaches the use of 5 to 200 ppm peracetic and or perpropionic acids for treatment of seedling culture medium prior to planting; specifically for the control of slime, algae or fungi on the culture medium. The teaching is limited to the use of C<sub>2</sub> and C<sub>3</sub> acids and has no application to growing plant tissue.

30 JP 07258005 A teaches the use of high levels (1000 ppm) of peracetic acid for controlling bacteria on rice. This application is intended only to effect disease control and not for hydroponic growth of the grain matter.

DE 3003875 A describes the use of C<sub>1</sub>–C<sub>4</sub> peracids and hydrogen peroxide to control phytopathogenic pests on soil. This reference does not disclose any direct 35 application to plants.

Brief Description of the Invention

We have found that a mixed peracid treatment composition can be used to protect growing plant tissue from the undesirable effects of microbial attack. The 5 mixed peracid materials used in this invention can be applied to growing plant tissues and can provide residual antimicrobial effects after the plant has completed its growth cycle, fruit or vegetable material have been harvested and sent to market. The materials of the invention have been found to be excellent antimicrobial compounds but pose little toxic effects to agricultural workers or the ultimate 10 consumer.

We have found that peroxy acid materials can be an effective treatment of living or growing plant tissues including seeds, roots, tubers, seedlings, cuttings, rooting stock, growing plants, produce, fruits and vegetables, etc. Under certain circumstances, a single peroxyacid material can be effective, however, in other 15 circumstances, a mixed peroxy acid has substantially improved and surprising properties.

The invention involves a peroxyacid antimicrobial concentrate and diluted end use composition including an effective microbicidal amount of a C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid such as peracetic acid, an effective microbicidal amount of a 20 C<sub>5</sub>–C<sub>12</sub> peroxyacid, preferably with a C<sub>6</sub>–C<sub>12</sub> or a C<sub>8</sub>–C<sub>12</sub> peroxy acid, or mixtures thereof. The concentrate composition can be diluted with a major proportion of water to form an antimicrobial sanitizing use solution having a pH in the range of about 2 to 8, with a C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid concentration of at least about 4 ppm, preferably about 10 to 75 ppm, and a C<sub>5</sub>–C<sub>12</sub>, a C<sub>6</sub>–C<sub>12</sub>, or a C<sub>8</sub>–C<sub>12</sub> 25 peroxyacid concentration of at least about 1 ppm, preferably about 1 to 25 ppm. Other components may be added such as a hydrotrope coupling agent for solubilizing the peroxyfatty acid in the concentrate form and when the concentrate composition is diluted with water.

The invention involves a method of controlling fungi and microbial plant 30 pathogens in growing plants by treating said growing plants with a dilute aqueous solution comprising an effective amount of a C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid and an aliphatic C<sub>5</sub>–C<sub>12</sub>, a C<sub>6</sub>–C<sub>12</sub> or a C<sub>8</sub>–C<sub>12</sub> peroxy carboxylic acid.

The invention further involves a process for controlling fungi and microbial plant pathogens in growing plants by diluting in an aqueous liquid a concentrate 35 containing: about 1 to 20 wt-% of a C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid; about 0.1 to 20

wt-% of an aliphatic C<sub>5</sub>–C<sub>12</sub>, a C<sub>6</sub>–C<sub>12</sub> or a C<sub>8</sub>–C<sub>12</sub> peroxy carboxylic acid to form a solution; and contacting said growing plants with said solution.

The invention further involves a process for controlling fungi and microbial plant pathogens in growing plants by diluting in an aqueous liquid a concentrate containing: about 1 to 20 wt-% of a C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid; about 0.1 to 20

5 wt-% of an aliphatic C<sub>5</sub>–C<sub>12</sub>, a C<sub>6</sub>–C<sub>12</sub> or a C<sub>8</sub>–C<sub>12</sub> peroxy carboxylic acid; about 5 to 40 wt-% of a C<sub>2</sub>–C<sub>4</sub> carboxylic acid; about 1 to 20 wt-% of an aliphatic C<sub>8</sub>–C<sub>12</sub> carboxylic acid; and about 1 to 30 wt-% of hydrogen peroxide to form a solution; and contacting said growing plants with said solution.

10 In contrast to the prior art, we have discovered that at a low pH, (e.g. preferably less than 7) C<sub>5+</sub> peroxy acids such as peroxyfatty acids are very potent biocides at low levels when used in combination with a C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid such as peroxyacetic acid, a synergistic effect is obtained, providing a much more potent biocide than can be obtained by using these components separately.

15 This means that substantially lower concentrations of biocide can be used to obtain equal biocidal effects.

As the term is used herein, a C<sub>5</sub>–C<sub>12</sub> peroxyacid (or peracid) is intended to mean the product of the oxidation of a C<sub>5</sub>–C<sub>12</sub> acid such as a fatty acid, or a mixture of acids, to form a peroxyacid having from about 5 to 12 carbon atoms per molecule.

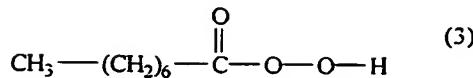
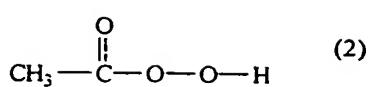
20 The C<sub>5</sub>–C<sub>12</sub> peroxy acids are preferably aliphatic (straight or branched).

Peroxy carboxylic acid is intended to mean the product of oxidation of a C<sub>2</sub>–C<sub>4</sub> carboxylic acid, or a mixture thereof. This includes both straight and branched C<sub>2</sub>–C<sub>4</sub> carboxylic acids.

25 The claimed invention includes a method of controlling fungi and microbial plant pathogens in growing plants. This treatment utilizes a combination of two different peroxy acids. This mixture comprises at least 4 parts per million (ppm) of a smaller C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid and at least 1 ppm of a larger C<sub>5</sub>–C<sub>12</sub> peroxy carboxylic acid. The preferred mixture comprises at least 4 ppm of a smaller C<sub>2</sub>–C<sub>4</sub> peroxy acid and at least 1 ppm of a large aliphatic C<sub>8</sub>–C<sub>12</sub> peroxy acid.

30 An especially preferred embodiment of the composition includes a mixture of peroxyacetic acid (given as formula (2)) and peroctanoic acid (given here as formula (3)).

6



5        The composition also may contain a hydrotrope for the purpose of increasing the aqueous solubility of various slightly soluble organic compounds. The preferred embodiment of the invention utilizes a hydrotrope chosen from the group of n-octanesulfonate, a xylene sulfonate, a naphthalene sulfonate, ethylhexyl sulfate, lauryl sulfate, an amine oxide, or a mixture thereof.

10      The composition may also contain a chelating agent for the purpose of removing ions from solution. The preferred embodiment of the invention uses 1-hydroxyethylidene-1,1-diphosphonic acid.

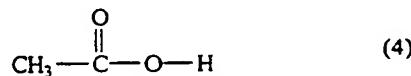
15      Further, the invention also provides a process of controlling fungi and microbial plant pathogens in growing plants. In this embodiment, the plant is contacted with a solution made by diluting in an aqueous liquid a concentrate comprising two peroxy acids. This mixture includes C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid and a larger C<sub>8</sub>–C<sub>12</sub> peroxy carboxylic acid. The preferred mixture includes about 1–20 weight percent (wt %) of a smaller C<sub>2</sub>–C<sub>4</sub> peroxy acid and about 0.1–20 wt % of a larger C<sub>8</sub>–C<sub>12</sub> peroxy acid. An especially preferred embodiment of the

20      composition includes a mixture of peroxyacetic acid and peroxyoctanoic acid. The composition may further contain about 1–15 wt % of a hydrotrope and about 5 wt-% of a chelating agent.

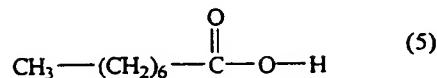
25      Finally, the invention also provides a process of controlling fungi and microbial plant pathogens in growing plants. In this embodiment, the plant is contacted with a solution made by diluting in an aqueous liquid a concentrate containing two peroxy acids. This mixture includes a smaller C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid and a larger C<sub>8</sub>–C<sub>12</sub> aliphatic peroxy carboxylic acid. An especially preferred embodiment of the composition includes a mixture of peroxyacetic acid and peroctanoic acid. The composition may further contain a hydrotrope and a

30      chelating agent. Further, the solution contains about 1–30 wt % of hydrogen

peroxide ( $H_2O_2$ ). The preferred composition includes a mixture of acetic acid (formula (4)) and octanoic acid (formula (5)).



5



### Detailed Description of the Invention

#### Peracids

10 We have found surprisingly that peroxy acid compounds of the invention can be contacted directly with living plant tissue in the form of a seed, a cutting, a root stock, graft, tuber juvenile or adult plant and reduced microbial populations without substantially affecting the health of the living tissue.

15 The invention is also based upon the surprising discovery that when a  $C_5-C_{12}$  peroxyacid is combined with a  $C_2-C_4$  peroxydicarboxylic acid, a synergistic effect is produced and greatly enhanced antimicrobial activity is exhibited when compared to the  $C_8-C_{12}$  peroxyacid or the  $C_2-C_4$  peroxydicarboxylic acid alone. The present blend of a  $C_8-C_{12}$  peroxyacid and a  $C_2-C_4$  peroxydicarboxylic acid can effectively kill microorganisms (e.g., a 5  $\log_{10}$  reduction in 30 seconds) from a concentration level, 20 below 100 ppm and as low as 20 ppm of the peracid blend.

25 A variety of  $C_5-C_{12}$  peroxyacids may be employed in the composition of the invention such as peroxyfatty acids, monoperoxy- or diperoxydicarboxylic acids, and peroxyaromatic acids. The  $C_5-C_{12}$  peroxyacids employed in the present invention may be structurally represented as:  $R_1-CO_3H$ , wherein  $R_1$  is a hydrocarbon moiety having from about 4 to 11 carbon atoms.  $R_1$  may have substituents in the chain, e.g.,  $-OH$ ,  $CO_2H$ , or heteroatoms (e.g.,  $-O-$  as in alkylether carboxylic acids), as long as the antimicrobial properties of the overall composition are not significantly affected. It should be recognized that " $R_1$ " substituents or heteroatoms may change the overall acidity (i.e.,  $pK_a$ ) of the carboxylic acids herein described. 30 Such modification is within the contemplation of the present invention provided the advantageous antimicrobial performance is maintained. Furthermore,  $R_1$  may be

linear, branched, cyclic or aromatic. Preferred hydrocarbon moieties (i.e. preferred R<sub>1</sub>'s) include linear, saturated, hydrocarbon aliphatic moieties having from 7 to 11 carbon atoms (or 8 to 12 carbon atoms per molecule).<sup>8</sup>

Specific examples of suitable C<sub>8</sub>–C<sub>12</sub> carboxylic fatty acids which can be reacted with hydrogen peroxide to form peroxyfatty acids include such saturated fatty acids as caprylic (octanoic) (C<sub>8</sub>), pelargonic (nonanoic) (C<sub>9</sub>), capric (decanoic) (C<sub>10</sub>), undecyclic (undecanoic) (C<sub>11</sub>), lauric (dodecanoic) (C<sub>12</sub>). These acids can be derived from both natural and synthetic sources. Natural sources include animal and vegetable fats or oils which should be fully hydrogenated. Synthetic acids can be produced by the oxidation of petroleum wax. Particularly preferred peroxyfatty acids for use in the composition of the invention are linear monoperoxy aliphatic fatty acids such as peroxyoctanoic acid, peroxydecanoic acid, or mixtures thereof.

Other suitable peroxyacids are derived from the oxidation of dicarboxylic acids and aromatic acids. Suitable dicarboxylic acids include sebacic acid (C<sub>10</sub>). An example of a suitable aromatic acid is benzoic acid. These acids can be reacted with hydrogen peroxide to form the peracid form suitable for use in the composition of the invention. Preferred peracids in this group include monoperoxy- or diperoxyadipic acid, monoperoxy- or diperoxysebacic acid, and peroxybenzoic acid.

The above peroxyacids provide antibacterial activity against a wide variety of microorganisms, such as gram positive (e.g., *Staphylococcus aureus*) and gram negative (e.g., *Escherichia coli*, *salmonella*, etc.) microorganisms, yeast, molds, bacterial spores, etc. When the above C<sub>5</sub>–C<sub>12</sub> peroxyacids are combined with a C<sub>2</sub>–C<sub>4</sub> peroxydicarboxylic acid, greatly enhanced activity is shown compared to the C<sub>2</sub>–C<sub>4</sub> peroxydicarboxylic acid alone or the C<sub>8</sub>–C<sub>12</sub> peroxyacid alone. The C<sub>2</sub>–C<sub>4</sub> peroxydicarboxylic acid component can be derived from a C<sub>2</sub>–C<sub>4</sub> carboxylic acid or dicarboxylic acid by reacting the acid, or the corresponding anhydride or acid chloride, with hydrogen peroxide. Examples of suitable C<sub>2</sub>–C<sub>4</sub> carboxylic acids include acetic acid, propionic acid, glycolic acid, and succinic acid or their corresponding anhydrides or acid chlorides. Preferable C<sub>2</sub>–C<sub>4</sub> peroxydicarboxylic acids for use in the composition of the invention include peroxyacetic acid, peroxypropionic acid, peroxyglycolic acid, peroxysuccinic acid, or mixtures thereof.

The antimicrobial concentrate of the present invention can comprise about 0.1 to 20 wt. %, preferably about 0.1 to 5 wt. %, and most preferably about 0.1 to 2 wt. % of a C<sub>8</sub>–C<sub>12</sub> peroxyacid, and about 1 to 20 wt. %, preferably about 1 to 15 wt. % and most preferably 4–15 wt. % of a C<sub>2</sub>–C<sub>4</sub> peroxydicarboxylic acid. The

concentrate composition preferably has a weight ratio of C<sub>2</sub>–C<sub>4</sub> peroxy carboxylic acid to C<sub>8</sub>–C<sub>12</sub> peroxyacid of about 15:1 to 1:1. The concentrate contains sufficient acid so that the end use solution has a pH of about 2 to 8, preferably about 3 to 7. Some acidity may come from an inert acidulant which may be optionally added (e.g., 5 sulfuric or phosphoric acid).

The peracid components used in the composition of the invention can be produced in a simple manner by mixing a hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) solution, or by utilizing powdered peroxide generators such as percarbonates or perborates, with the desired amount of acid. With the higher molecular weight fatty acids, a hydrotrope 10 coupler may be required to help solubilize the fatty acid. The H<sub>2</sub>O<sub>2</sub> solution also can be added to previously made peracids such as peracetic acid or various perfatty acids to produce the peracid composition of the invention. The concentrate can contain about 1 to 30 wt. %, preferably about 5 to 25 wt. % of hydrogen peroxide.

The concentrate composition can further comprise a free C<sub>8</sub>–C<sub>12</sub> carboxylic acid, a free C<sub>2</sub>–C<sub>4</sub> carboxylic acid, or mixtures thereof. The free acids will 15 preferably correspond to the starting materials used in the preparation of the peroxyacid components. The free C<sub>8</sub>–C<sub>12</sub> carboxylic acid is preferably linear and saturated, has 8 to 12 carbon atoms per molecule, and can also comprise a mixture of acids. The free C<sub>8</sub>–C<sub>12</sub> carboxylic acid and free C<sub>2</sub>–C<sub>4</sub> carboxylic acid can be 20 present as a result of an equilibrium reaction with the hydrogen peroxide to form the peroxyacids.

#### Other Components

Various optional materials may be added to the composition of the invention 25 to help solubilize the fatty acids, restrict or enhance the formation of foam, to control hard water, to stabilize the composition, or to further enhance the antimicrobial activity of the composition.

The composition of the invention can contain a surfactant hydrotrope coupling agent or solubilizer that permits blending short chain perfatty acids in 30 aqueous liquids. Functionally speaking, the suitable couplers which can be employed are non-toxic and retain the fatty acid and the perfatty acid in aqueous solution throughout the temperature range and concentration to which a concentrate or any use solution is exposed.

Any hydrotrope coupler may be used provided it does not react with the other 35 components of the composition or negatively affect the antimicrobial properties of

10 the composition. Representative classes of hydrotropic coupling agents or solubilizers which can be employed include anionic surfactants such as alkyl sulfates and alkane sulfonates, linear alkyl benzene or naphthalene sulfonates, secondary alkane sulfonates, alkyl ether sulfates or sulfonates, alkyl phosphates or 5 phosphonates, dialkyl sulfosuccinic acid esters, sugar esters (e.g., sorbitan esters), amine oxides (mono-, di-, or tri-alkyl) and C<sub>8</sub>-C<sub>10</sub> alkyl glucosides. Preferred coupling agents for use in the present invention include n-octanesulfonate, available as NAS 8D from Ecolab, n-octyl dimethylamine oxide, and the commonly available aromatic sulfonates such as the alkyl benzene sulfonates (e.g. xylene sulfonates) or 10 naphthalene sulfonates.

Some of the above hydrotropic coupling agents independently exhibit antimicrobial activity at low pH. This adds to the efficacy of the present invention, but is not the primary criterion used in selecting an appropriate coupling agent. Since it is the presence of perfatty acid in the protonated neutral state which provides 15 biocidal activity, the coupling agent should be selected not for its independent antimicrobial activity but for its ability to provide effective interaction between the substantially insoluble perfatty acids described herein and the microorganisms which the present compositions control.

The hydrotropic coupling agent can comprise about 0.1 to 30 wt. %, 20 preferably about 1 to 15 wt. %, and most preferably about 2 to 15 wt. % of the concentrate composition.

Compounds such as mono, di and trialkyl phosphate esters may be added to the composition to suppress foam. Such phosphate esters would generally be produced from aliphatic linear alcohols, there being from 8 to 12 carbon atoms in the 25 aliphatic portions of the alkyl phosphate esters. Alkyl phosphate esters possess some antimicrobial activity in their own right under the conditions of the present invention. This antimicrobial activity also tends to add to the overall antimicrobial activity of the present compositions even though the phosphate esters may be added for other reasons. Furthermore, the addition of nonionic surfactants would tend to 30 reduce foam formation herein. Such materials tend to enhance performance of the other components of the composition, particularly useful nonionic surfactant for use as a defoamer is nonylphenol having an average of 12 moles of ethylene oxide condensed thereon, it being encapped with a hydrophobic portion comprising an average of 30 moles of propylene oxide.

Chelating agents can be added to the composition of the invention to enhance biological activity, cleaning performance and stability of the peroxyacids. For example, 1-hydroxyethylidene-1, 1-diphosphonic acid commercially available from the Monsanto Company under the designation "DEQUEST" has been found to 5 be effective. Chelating agents can be added to the present composition to control or sequester hardness ions such as calcium and magnesium. In this manner both detergency and sanitization capability can be enhanced.

Other materials which are sufficiently stable at the low pH contemplated by the present composition may be added to the composition to impart desirable 10 qualities depending upon the intended ultimate use. For example, phosphoric acid ( $H_3PO_4$ ) can be added to the composition of the invention. Additional compounds can be added to the concentrate (and thus ultimately to the use solution) to change its color or odor, to adjust its viscosity, to enhance its thermal (i.e., freeze-thaw) stability or to provide other qualities which tend to make it more marketable.

15 The composition of the invention can be made by combining by simply mixing an effective amount of a  $C_8-C_{12}$  peroxyacid such as a peroxyfatty acid with some source of a  $C_2-C_4$  peroxycarboxylic acid such as peroxyacetic acid. This composition would be formulated with preformed perfatty acid and preformed peroxyacetic acid. A preferred composition of the invention can be made by mixing 20 a  $C_2-C_4$  carboxylic acid, a  $C_8-C_{12}$  carboxylic acid, a coupler and a stabilizer and reacting this mixture with hydrogen peroxide. A stable equilibrium mixture is produced containing a  $C_2-C_4$  peroxycarboxylic acid and a  $C_8-C_{12}$  peroxyacid by allowing the mixture to stand for from one to seven days at 15°C. to 25°C. As with any aqueous reaction of hydrogen peroxide with a free carboxylic acid, this gives a 25 true equilibrium mixture. In this case, the equilibrium mixture will contain hydrogen peroxide, a  $C_2-C_4$  peroxycarboxylic acid, a  $C_8-C_{12}$  carboxylic acid, a  $C_2-C_4$  peroxycarboxylic acid, a  $C_8-C_{12}$  peroxyacid, water, and various couplers and stabilizers.

30 By using the above approach, the composition of the invention can be formulated by merely mixing readily available raw materials, e.g., acetic acid, hydrogen peroxide and fatty acid. By allowing solution time for equilibrium to be obtained, the product containing both of the active biocides is obtained. In varying the ratio of  $C_2-C_4$  carboxylic acid to  $C_8-C_{12}$  carboxylic acid, it is easy to vary the ratio of  $C_2-C_4$  peroxycarboxylic acid to  $C_8-C_{12}$  peroxyacid.

## 12 Method of Treatment

The present invention contemplates a concentrate composition which is diluted to a use solution prior to its utilization as a microbicide. Primarily for reasons of economics, the concentrate would normally be marketed and the end user 5 would dilute the concentrate with water to a use solution. A preferred antimicrobial concentrate composition comprises about 0.1 to 20 wt. %, preferably about 0.1 to 5 wt. %, of a C<sub>8</sub>–C<sub>12</sub> peroxyfatty acid, about 1 to 20 wt. % of a C<sub>2</sub>–C<sub>4</sub> peroxycarboxylic acid, about 1 to 15 wt. % of a hydrotrope coupling agent, and about 1 to 30 wt. % of hydrogen peroxide. Other acidulants may optionally be 10 employed in the composition such as phosphoric acid.

The level of active components in the concentrate composition is dependent upon the intended dilution factor and desired acidity in the use solution. The C<sub>8</sub>–C<sub>12</sub> peroxyacid component is generally obtained by reacting a C<sub>8</sub>–C<sub>12</sub> carboxylic acid with hydrogen peroxide in the presence of a C<sub>2</sub>–C<sub>4</sub> carboxylic acid. The resulting concentrate is diluted with water to provide the use solution. Generally, a dilution of 1 fluid oz. to 4 gallons (i.e. dilution of 1 to 500 by volume) of water can be obtained with 2% to 20% total peracids in the concentrate.

The compositions of the invention can be applied to growing plant tissue in a variety of techniques. The aqueous solution can be sprayed, painted, daubed, fogged, flooded onto or into the plant, the plant hydroponic substrate, the agricultural earth. The material can be reapplied periodically as needed.

### Examples

25 **Initial testing**

The examples given herein describe the treatment of a snow mold fungus isolated from a mountain ash (*Sorbus Americana*) tree. Testing consisted of an untreated control versus treating the fungus with peroxy acetic acid (POAA) ( $C_2$ ) and peroxyacetic/peroxyoctanoic (POAA/POOA) ( $C_2/C_8$ ) composition. This testing showed that the latter was very effective in killing the fungus.

Thus, approximately 10 cm<sup>2</sup> of "snow mold" was removed from the branch section and split it into three parts; putting each into water. One section was untreated. A second was treated with 500 ppm peroxyacetic acid, and a third was treated using a mixed peracid system (C<sub>2</sub>/C<sub>8</sub>). After two days:

1. The control was still thriving and very moist to the touch; with a large jelly-like volume of fungi.
2. The POAA sample treated was approximately 1/2 dead with what appears to be a dead outer crust over approximately 2/3 the sample and a dried 5 completely dead patch over the rest. The 2/3 crusted area still had the soft jelly-like mass.
3. The C2/C8 penoxyacid blend treated Fungi appeared all dead and dried up. No mass of jelly left.

10

#### Peroxy Acetic Acids Effect On Growing Plant Tissue

The study was conducted on growing plant tissues using sprays done before 8:00 AM. Testing was conducted on usually disease free plants to investigate possible side effects such as burning of plant tissue. The data demonstrates both peracid compositions to be similar in effect to the growing plant tissue during 15 application; with both being relatively inert on the tissue surfaces except for high surface area crops such as Asparagus leaves (no effect on stems).

Peracid Formula	Dosed Peracid (ppm) <sup>3</sup>	Plant type/treated area	Observed effects 8 hrs after application
POAA <sup>1</sup>	100	American Basswood/lower leaves	No changes, no burning or tissue damage
POAA/POOA <sup>2</sup>	100	American Basswood/lower leaves	No changes, no burning or tissue damage
POAA <sup>1</sup>	100	American Basswood/lower leaves (approx. 30')	No changes, no burning or tissue damage
POAA/POOA <sup>2</sup>	100	American Basswood/lower leaves (approx. 30')	No changes, no burning or tissue damage
POAA <sup>1</sup>	100	State Fair Apple/whole tree (approx. 7')	No changes, no burning or tissue damage
POAA/POOA <sup>2</sup>	100	State Fair Apple/whole tree (approx. 7')	No changes, no burning or tissue damage
POAA <sup>1</sup> POAA/POOA <sup>2</sup>	100 100	Wild rose bush/whole (3') Wild rose bush/whole (2')	No changes, no burning or tissue damage
POAA <sup>1</sup>	100	Asparagus/1/2 plant (approx. 4')	Some yellowing found, apparent leaf burn on treated halves
POAA/POOA <sup>2</sup>	100	Asparagus/1/2 plant (approx. 4')	Some yellowing found, apparent leaf burn on treated halves
POAA <sup>1</sup>	100	Pixwell Gooseberry/1/2 plant (approx. 3')	No change
POAA/POOA <sup>2</sup>	100	Pixwell Gooseberry/1/2 plant (approx. 3')	No change

1. POAA = peroxyacetic acid (C<sub>2</sub>).

2. POAA/POOA = peroxyacetic/peroxyoctanoic acids (C<sub>2</sub>/C<sub>8</sub>).

20 3. Total active peracid.

## Continued Tests For Peracids On Plant Tissue

Peracid <sup>A</sup>	Dosed ppm	Plant/Area <sup>B</sup>	Comments
1	100	Basswood/lower branches	No effects noted, no burn or bleaching
2	100	Basswood/lower branches	No effects note, no burn or bleaching
1	100	Rose/whole	No effects noted, no burn or bleaching
2	100	Rose/whole	No effects noted, no burn or bleaching
1	100	1/2 asparagus plant	Leaf yellowing and more bleaching – stopping application to avoid plant death
2	100	1/2 asparagus plant	Leaf yellowing and more bleaching – stopping application to avoid plant
1	100	Beacon Apple/whole State Fair Apple/whole	No effect, as above
2	100	State Fair Apple/whole NOTE: switched application order vs. last week	No effect, as above
1	100	Gooseberry/half	No effect, as above
2	100	Gooseberry/half	No effect, as above
1	100	Pontiac Potato Plant/whole	No effect, as above
2	100	Pontiac Potato Plant/whole	No effect, as above
1	100	Beta Grapevine/1/2 vine	No effect, as above
2	100	Beta Grapevine/1/2 vine	No effect, as above
Increased Dosage Tests:			
1	300	Wild Plum/approx. 1/2 tree	No effect, as above
2	300	Wild Plum/approx. 1/2 tree	No effect, as above
1	300	Pontiac Potato/whole	No effect, as above
2	300	Pontiac Potato/whole	No effect, as above

A. As per previous table

B. Same test plants and test area as per previous table, unless noted.

5

Peracid	Dosed ppm	Plant/Area	Comments
1	300	Approx. 10' Wild Plum/whole	No effect noticed, no burning or bleaching
2	300	Approx. 8' Wild Plum/whole	No effect noticed, no burning or bleaching
1	300	Red Pontiac/whole	Same as above – no effect
2	300	Red Pontiac/whole	Same as above no effect
1	300	Gooseberry/1/2	Same as above – no effect
2	300	Gooseberry/1/2	Same as above no effect
1	300	Beebalm (menarda)/10 plants	Slight leaf yellowing – killed snow mold on all plants – (approx. 16 with mold before treatment)
2	300	Beebalm (menarda)/10 plants	Slight leaf yellowing – killed snow mold on all plants – (approx. 16 with mold before treatment)
1	300	Rose/whole	No effects, as above
2	300	Rose/whole	No effects, as above
1	300	Beacon Apple/whole	No effects, as above
2	300	State Fair Apple. whole	No effects, as above

Peracid	Dosed ppm	Plant/Area	Comments
1	300	American Basswood/bottom	No effects, as above
2	300	American Basswood/bottom	No effects, as above
1	300	Hawthorne (approx. 7 <sup>1</sup> )/whole	No effects, as above
2	300	Hawthorne (approx. 8 1/2 <sup>1</sup> )/whole	No effects, as above
1	300	Betterboy Tomato/whole	No effects, as above
2	300	Betterboy Tomato/whole	No effects, as above
1	300	Beta Grape/ 1/2 (10') vine	No effects, as above
1	300	Beta Grape/ 1/2 (10') vine	No effects, as above

### Working Examples

### Working Example 1:

5 **Minimum Inhibitory Concentration of Peracetic Acid vs. Mixed  
Peracid Composition against *Botrytis cinerea* Plant Pathogen**

This example compares the effect of the prior art using peracetic acid (POAA) vs. the combination peracid formula of the present invention. The objective is to determine the minimum inhibitory concentrations against the *Botrytis cinerea* ATCC 11542 plant pathogen organism.

*Botrytis cinerea* culture was prepared by inoculating the center of ten Sabouraud Dextrose Agar filled dishes and incubating at 26°C–30° for 15 days. Mycelial mats were removed by adding 10 mL of sterile water and using a sterile 15 spatula to brush the microbial growth from agar surface. The suspension was transferred to tissue grinder and macerated with 10–25 mL of sterile water, then filtered through cheesecloth and stored in a glass bottle at 4°C until test time.

Product dilutions were prepared in a Sabouraud Dextrose broth at levels of % POAA adjusted to deliver a part per million consternation of 30, 45, 60, 75, 150 and 300. Solutions were inoculated with 0.5 mL of the prepared culture suspensions and incubated at 26°C–30°C for 15 days to observe growth. One tube of Sabouraud Dextrose broth was used as a positive growth control for each culture, and 1 tube was used to observe sterility of broth.

Table 1 compares the minimum inhibitory concentration (the lowest peracid efficacy level to impart no growth in the plant pathogen) of the mixed peracid compositions (line 1) and those of the single peracid (lines 2) for the reduction of

16 common plant pathogens. The results demonstrate the 5-fold improved efficacy results of the mixed peracid compositions for the reduction of *Botrytis cinera* ATCC 11542 (cf., lines 1 and 2); i.e., the minimum inhibitory concentration for controlling *Botrytis cinera* is 60 ppm peracid with the current formulas, while 300 ppm peracid 5 is required using the prior art.

Table 1

10 **Minimum Inhibitory Concentration of Peracetic Acid vs. a Novel Mixed Peracid Composition against *Botrytis cinera* Plant Pathogen**

Run #	Peracid Test	Organism	Minimum Inhibitory Concentration					
			30 (ppm)	45 (ppm)	60 (ppm)	75 (ppm)	150 (ppm)	300 (ppm)
<b>current art:</b>								
1	Mixed Peracid Formula (POAA/POOA) <sub>1</sub>	<i>Botrytis cinera</i>	+	+	-	-	-	-
<b>prior art:</b>								
2	Prior Art (POAA) <sup>2</sup>	<i>Botrytis cinera</i>	+	+	+	+	+	-

(-) = no growth, (+) = growth

15 1 POAA/POOA = peroxyacetic/peroxyoctanoic acids

2 prior art as demonstrated in GB 2187958A.

3 POAA = peroxyaceticacid

20

**Working Example 2:**

Table 2 compares the antimicrobial effect of using the present mixed peracid system and prior known materials as fruit and vegetable treatments for human pathogen reduction. Thus, we compare the novel mixed C<sub>2</sub>/C<sub>8</sub> peracid system to of 25 using sodium hypochlorite or peracetic acid.

Testing was performed on three produce surfaces: tomatoes, leaf lettuce and apples, and also using four test organisms – *Listeria monocytogenes*, *Salmonella javiana*, natural bacteria flora, and *penicillium expansum*. For the pathogenic organisms, a 1:10 dilution of a 10<sup>7</sup> CFU/mL test system suspension was prepared. 30 50 grams of the leaf lettuce, a whole tomato, or whole apple were placed into a plastic bag. Each bag was inoculated with 10.0 mL test system suspension resulting in an inoculum level of 10<sup>7</sup> CFU/mL (colony forming units per milliliter). Bags were gently shaken for even distribution of the test system for 5 minutes. The produce types were then stored overnight at 4°C. Untreated controls (no bacterial

17

inoculum-10 mL of phosphate buffered dilution water) were also prepared. Two liter volumes of the test solutions were prepared in 4 liter beakers. Solutions were prepared in laboratory tap water to simulate industry conditions. The vegetables were then exposed to the test solution by submersion for 5 minutes at 72°F. At the 5 end of the specified exposure time, the vegetables were removed from the test solution and thoroughly rinsed under fresh running tap water. Fifty grams of the vegetable (lettuce) or whole produce (tomato or apple) + 100 mL buffered dilution water were placed into a stomacher bag. The vegetables were stomached (lettuce) or massaged (tomato or apple) for 60 seconds. Serial dilutions were made and plated 10 on TGE (*Salmonella*), SAB (*penicillium*) and BHI (*Listeria*). Plates were incubated at 35°C for 48 hours. The following controls were performed in each test: Untreated control (no inoculation, no chemical treatment) for the background microbial load, an inoculated control (inoculated but no chemical treatment) for the inoculum plus background microbial load on vegetable surface, and a tap Water control with a 15 serial dilution to determine if microbial contamination was present in the rinse water or product diluent.

The results demonstrate the enhanced effectiveness of the mixed peracid system vs. the conventional (sodium hypochlorite or peracetic acid) treatment systems. Thus, substantial improvements over the sodium hypochlorite treatments 20 are found when using the peracid formulas. Additionally, the mixed peracid composition of the present invention (C<sub>2</sub>/C<sub>8</sub>) yields log reduction improvements of 0.5-1.0 log vs. a single C<sub>2</sub> peracid composition; or comparable log reductions using 50% less of the active peracids.

Table 2

## 5 The Comparative Effect Of A Mixed Peracid System vs. Prior Art For Effecting Microbial Control On Harvested Fruit And Vegetables

Test #	Treated Produce & Test Organism	Treatment Process	Active Treatment Concentration (ppm)	Average Microbial Count (CFU/mL)	Log Reduction
1	<b>Tomatoes</b> <i>L. monocytogenes</i>	inoculated control	0 (control)	$4.8 \times 10^5$	0 <sup>1</sup>
2	<i>L. monocytogenes</i>	peracetic acid <sup>1</sup>	50 ppm POAA <sup>1</sup>	$1.4 \times 10^5$	0.5
3	<i>L. monocytogenes</i>	novel mixed peracids <sup>2</sup>	50 ppm (POAA+POOA) <sup>2</sup>	$2.4 \times 10^4$	1.3
4	<i>S. javiana</i>	inoculated control	0 (control)	$1.6 \times 10^6$	0 <sup>1</sup>
5	<i>S. javiana</i>	sodium hypochlorite <sup>3</sup>	25 ppm NaOCl <sup>3</sup>	$7.4 \times 10^5$	0.3
6	<i>S. javiana</i>	peracetic acid <sup>1</sup>	25 ppm POAA <sup>1</sup>	$1.7 \times 10^4$	2.0
7	<i>S. javiana</i>	peracetic acid <sup>1</sup>	50 ppm POAA <sup>1</sup>	$8.1 \times 10^3$	2.3
8	<i>S. javiana</i>	peracetic acid <sup>1</sup>	75 ppm POAA <sup>1</sup>	$4.6 \times 10^3$	2.5
9	<i>S. javiana</i>	novel mixed peracids <sup>2</sup>	50 ppm (POAA+POOA) <sup>2</sup>	$2.3 \times 10^3$	2.8
10	<b>Lettuce</b> natural flora	untreated control <sup>1</sup>	0 (control)	$1.3 \times 10^8$	0 <sup>1</sup>
11	natural flora	sodium hypochlorite <sup>3</sup>	100 ppm NaOCl <sup>3</sup>	$1.4 \times 10^8$	0.2
12	natural flora	peracetic acid <sup>1</sup>	75 ppm POAA <sup>1</sup>	$2.1 \times 10^7$	1.0
13	natural flora	novel mixed peracids <sup>2</sup>	50 ppm (POAA+POOA) <sup>2</sup>	$1.8 \times 10^7$	1.1
14	<b>Apple</b> <i>P. expansum</i>	inoculated control	0 (control)	$2.8 \times 10^4$	0 <sup>1</sup>
15	<i>P. expansum</i>	peracetic acid <sup>1</sup>	80 ppm POAA <sup>1</sup>	$7.5 \times 10^3$	0.6
16	<i>P. expansum</i>	novel mixed peracids <sup>2</sup>	80 ppm (POAA+POOA) <sup>2</sup>	$1.4 \times 10^3$	1.3

10      1 per prior art GB 2187958A

1 No treatment control

2 present invention

3 chlorine washes

19  
**Working Example 3:**

**Temperature Effects For Mixed Peracid Compositions**

Example 3 compares the temperature effect of using peracetic acid (POAA) vs. a mixed peracid system under cold water applications. The experiments were run 5 as in example 2; however, a cold water (40°F) application temperature was used.

In contrast to the experiments of Table 2 where peracetic acid applications outperformed sodium hypochlorite for surface reduction of microbials, the results of Table 3 demonstrate that colder treatment temperatures impedes the activity of the single peracid composition. Conversely, the mixed peracid system is unexpectedly 10 less effected, and still substantially outperforms the known systems.

**Table 3**  
**Temperature Evaluation For Peracid Treatment of Tomatoes**

Run #	Treatment Condition (all at 40°F)	Peracid Treatment Level (ppm)	Log Reduction (of <i>S. javiana</i> ) on Tomatoes
prior art examples:			
1	Sodium Hypochlorite	80 ppm	1.4
2	Peracetic Acid	80 ppm	0.0
current art example:			
3	Peracetic-Peroctanoic Acid Mix	80 ppm	2.3

**Working Example 4:**

15      **Comparative Peracid Treatment of a Substrate vs. Alfalfa Sprouts For  
Microbial Control During Hydroponic Growing**

The objective of this example was to compare the use of a single peracid vs. mixed peracids for microbial reduction during hydroponic growth of alfalfa sprouts. Ongoing concern in the industry is the control of microbial populations; especially 20 human and plant patogenics, but also nutrition-solution molds and fungi. The following test was conducted to determine potential microbial control during the hydroponic growing cycle.

Table 4 compares the results of utilizing the continuous hydroponic treatment technique of the present invention in contrast to U.S. 5,168,655 which utilizes 25 peracetic acid disinfection of hydroponic substrates; i.e., substantial (>5-log) microbial reductions can be found if the peracid treatment is ongoing vs. essentially no reduction if only the substrate is treated to hydroponic growth.

**Table 4**  
**Peracetic Acid (POAA) and Peracetic-Peroctanoic (POAA-POOA) Treatments of Alfalfa Sprouts comparing The Prior and Current Art**

Aerobic Plate Count Results						
	1	2	3	4	5	6
Peracid Treatment Condition	Peracid Concentration (ppm)	Day 1	Day 2	Day 3	Day 4	
<b>Control Study</b>						
<b>1 Water Control (no peracid treatment)<sup>2</sup></b>	0	0	0	0	0	0
<b>Current Art Examples</b>						
<b>6 POAA-POOA treated<sup>3</sup></b>	40 ppm	1.5	<0.1	no reduction	no reduction	
7	80 ppm	1.8	1.9	0.5	0.2	
8	160 ppm	1.9	5.4	3.1	2.9	
9	320 ppm	1.6	5.9	5.0	2.5	
<b>Prior Art Examples</b>						
<b>9 POAA treated substrate<sup>4</sup></b>	40 ppm	<0.5	no	no	no	
10	80 ppm	<0.5	reduction	reduction	reduction	
11	160 ppm	<0.5	at any level	at any level	at any level	
	320 ppm	<0.5				

- 1) Microbial reduction vs. the water control study without any type of peracid treatment
- 2) The water control was used as the background basis for the treatment efficiency Log reductions for each treatment day. Typically log counts of  $\sim 1 \times 10^9$  cfu/ml were found.
- 3) POAA-POOA = peroxyacetic acid from an equilibrium composition, using the method of application of US 5,168,655.
- 4) POAA = peroxyacetic-peroxyoctanoic acids from an equilibrium composition, using the current art method of application.

**Working Example 5:****POAA and POAA-POOA Treatment of Alfalfa Sprouts with Variable Concentrations of Peracids**

5        The objective of this example was to evaluate microbial reduction using peracid misting during the daily hydroponic growth of alfalfa sprouts; against natural bacterial flora. Commercially, bean and alfalfa sprouts are grown by overhead misting of seed plates for 3-5 days. The sprouts are harvested and the seed waste disposed of. An ongoing concern in the industry is the control of microbial  
10      populations; especially human and plant pathogenics, but also nutrition-solution molds and fungi. The following test was conducted to determine potential microbial control during the hydroponic growing cycle.

15      The alfalfa sprouts were soaked in various concentrations of equilibrium obtained peracetic (POAA) or peracetic-perooctanoic POAA-POOA solutions. One sample was soaked in water as a control. The following morning the alfalfa sprouts were placed into a sterile petri dish by evenly spreading the seeds on the bottom of the dish. The petri dishes were covered with cheese cloth for the growing procedure.

20      During growing (days 1-4) the alfalfa seeds were treated twice daily at 8:00 AM and 4:45 PM by misting with 10 ml of the same concentration of peracid in which they were soaked. The water control was misted with water. Microbial samplings were taken at 8:00 AM each of the 4 treatment days. A 1:10, wt:wt, mixture of sprout:water was stomached and plated onto TGE Agar subculture media using a Pour Plate Technique with phosphate buffered dilutions of  $10^{-3}$ ,  $10^{-5}$ ,  $10^{-7}$ . After 48 hours at 35°C the micro results were determined and are shown in Table 2:

25      The results of Table 5 demonstrate the ability to effect microbial populations during hydroponic growth of plant tissue using continuous peracid applications. In contrast to U.S. 5,168,655 which utilizes peracetic acid disinfection of hydroponic substrates prior to a crop production cycle, the microbial peracids of the present invention demonstrates the novel utility of using peracids to effect continuous  
30      microbial control during the entire hydroponic growth cycle, without loss to crop yield (see examples 4 and 5). The data also indicates the need to modify the dosing procedure to enhance the microbial reduction near the end of the hydroponic cycle. This hypothesis is tested in Example 3.

**Table 5**  
**Peracetic Acid (POAA) and Peroxyacetic-Peroctanoic (POAA-POOA) Treatments of Alfalfa Sprouts Using Variable Treatment Concentrations**

**Aerobic Plate Count Results**

	1	2	3	4	5	6
Peracid Treatment Condition	Peracid Concentration (ppm)	Day 1	Day 2	Day 3	Day 4	Microbial Reduction
1 Water Control (no peracid treatment) <sup>a</sup>	0	0	0	0	0	0
2 POAA treated <sup>b</sup>	40 ppm	0.4	0.1	no reduction	no reduction	0
3	80 ppm	1.7	0.5	0.2	0.1	0.1
4	160 ppm	1.9	2.2	0.4	0.2	0.2
5	320 ppm	1.7	4.7	4.2	0.3	0.3
6 POAA-POOA treated <sup>c</sup>	40 ppm	1.5	<0.1	no reduction	no reduction	0
7	80 ppm	1.8	1.9	0.5	0.2	0.2
8	160 ppm	1.9	5.4	3.1	2.9	2.9
9	320 ppm	1.6	5.9	5.0	2.5	2.5

a) The water control was used as the background basis for the treatment efficiency Log reductions for each treatment day. Typically  $10^9$  cfu/ml.

b) POAA = peroxyacetic acid from an equilibrium composition.

c) POAA-POOA = peroxyacetic-peroxyoctanoic acids from an equilibrium composition.

**Working Example 6:****POAA and POAA-POOA Treatment of Alfalfa Sprouts with an Alternative Misting Procedure**

The objective of this example was to evaluate microbial reduction using a 5 more continuous (hourly) peracid misting procedure during the daily hydroponic growth of alfalfa sprouts; against natural bacterial flora. This should allow for a lower dosage profile of peracids.

Using the above microbial technique, the alfalfa sprouts were soaked in 80 ppm of POAA or POAA-POOA solutions. One sample was soaked in water as a 10 control. The following morning the alfalfa sprouts were placed into a sterile petri dish by evenly spreading the seeds on the bottom of the dish. The petri dishes were covered with cheese cloth for the growing procedure.

During growing (days 1-4) the alfalfa seeds were treated from 8:00 AM to 3:00 PM on an hourly basis by misting with 10 ml of the same concentration of 15 peracid in which they were soaked. The water control was misted with water. Microbial samplings were taken at 4:00 PM each of the 4 treatment days.

The results of Table 6 demonstrate the improved efficacy of: using a more continuous dosing system, and for using a mixed peracid system for microbial control during hydroponic growth of plant tissue. While peracetic acid can impart an 20 initial (day 1-2) microbial reduction vs. the control (cf., experiment 1,2), it fails after the 2nd growing day. Conversely the mixed peracetic-perooctanoic acid (POAA-POOA) system yields continuous microbial control (> 20log) over the entire sprout growing time (experiment 3). For all the experiments the germination rate of the seed was greater than 95%.

Table 6

Peracetic Acid (POAA) and Peracetic-Peroctanoic (POAA-POOA) Treatments of Alfalfa Sprouts Using A Seven Hour Per Day Misting Procedure

## Aerobic Plate Count Results

Sample Identification	CFU/ml	Day 1		Day 2		Day 3		Day 4	
		Log R <sup>a</sup>	CFU/ml	Log R <sup>a</sup>	CFU/ml	Log R <sup>a</sup>	CFU/ml	Log R <sup>a</sup>	CFU/ml
1 Water Control	2.0 x 10 <sup>8</sup>	0 <sup>b</sup>	1.1 x 10 <sup>8</sup>	0 <sup>b</sup>	7.8 x 10 <sup>8</sup>	0 <sup>b</sup>	2.1 x 10 <sup>9</sup>		
2 peracetic acid (POAA) 80 ppm	2.5 x 10 <sup>8</sup>	1.9	3.1 x 10 <sup>8</sup>	0.6	8.7 x 10 <sup>8</sup>	no red <sup>c</sup>	2.1 x 10 <sup>9</sup>	no red	
3 peracetic-peroctanoic acids (POAA-POOA) 80 ppm	4.0 x 10 <sup>7</sup>	3.7	8.0 x 10 <sup>6</sup>	2.1	6.5 x 10 <sup>6</sup>	2.1	1.3 x 10 <sup>7</sup>	2.2	

- a) Log R = log reduction.
- b) The water control was used as the background basis for the treatment efficiency Log reductions.
- c) no red = no reduction of the microbial load within the testing error range.

**Working Example 7:**

**POAA and POAA-POOA Continuous Misting Treatments of Alfalfa Sprouts**

The objective of this example was to evaluate microbial reduction using a 5 continuous (hourly over 24-hours) peracid misting procedure during the daily hydroponic growth of alfalfa sprouts: against natural bacterial flora.

Using the above microbial technique, the alfalfa sprouts were soaked in 80 ppm or POAA or POAA-POOA solutions. One sample was soaked in water as a control. The following morning the alfalfa sprouts were placed into a sterile petri 10 dish by evenly spreading the seeds on the bottom of the dish. The petri dishes were covered with cheese cloth or the growing procedure.

During growing (days 104) the alfalfa seeds were treated on an hourly basis (over the entire 24 hour day) by misting with 10 ml of the same concentration of peracid in which they were soaked. The water control was misted with water.

15 Microbial samplings were taken at 4:00 PM each of the 4 treatment days.

The results of Table 7 demonstrates the unexpected result that continuous 24 hour per day misting does not improve – over the previous examples 7-times per day peracid application – the final growing day microbial control efficacy for either peracid system during hydroponic production of plant tissue (cf., Table 6, 20 experiments 2.3 vs. Table 7 experiment 2.3). Additionally, the germination yield dropped off tremendously (< 50 %) in all the 24-hour per day studies as compared to the results of example 3.

**Table 7**  
**Peracetic Acid (POAA) and Peracetic-Peroctanoic (POAA-POOA) Treatments**  
**25 of Alfalfa Sprouts Using a Twenty Four Hour Per Day Misting Procedure**

**Aerobic Plate Count Results**

	Sample Identification	1	2	3
		Day 4 results		
		Microbial CFU/mL	Log R <sup>a</sup>	Germination rate <sup>b</sup>
1	Water Control	$1.0 \times 10^5$	0 <sup>c</sup>	< 50 %
2	peracetic acid (POAA) 80 ppm	$6.0 \times 10$	0.2	< 50 %
3	peracetic-peroctanoic acids (POAA-POOA) 80 ppm	$3.5 \times 10^6$	1.9	< 40 %

a) Log R = log reduction.

- b) A visual evaluation of % of seeds growing vs. non sprouted.
- c) The water control was used as the background basis for the treatment efficiency Log reductions.

5

### Working Example 8:

#### POAA and POAA-POOA Daily Misting Germination Rates of Alfalfa Sprouts

The objective of this example was to evaluate the germination rate for the various peracid applications during the hydroponic growing cycle of alfalfa sprouts.

Using the above microbial technique, the alfalfa sprouts were soaked in 80 ppm of POAA or POAA-POOA solutions. One sample was soaked in water as a control. The following morning the alfalfa sprouts were placed into a sterile petri dish by evenly spreading the seeds on the bottom of the dish. The petri dishes were covered with cheese cloth for the growing procedure.

During growing (days 1-4) the alfalfa seeds were treated using various application times as per examples 2-4 by misting with 10 ml of the same concentration of peracid in which they were soaked. The water control was misted with water. After 4 days the germination rate was visually determined.

The results of Table 8 demonstrate: that proper selection of the application rate and peracid composition is necessary to impart both microbial reduction and hydroponic seed germination. Both upper and lower application limits are found.

**Table 8**  
**Peracetic Acid (POAA) and Peracetic-Peroctanoic (POAA-POOA) Treatments**  
**of Alfalfa Sprouts Using Variable Treatment rates**

25

#### Aerobic Plate Count Results

	Peracid Treatment Rate	1	2	3
		Day 4 Microbial and Germination Results (Log Reduction and %)*		
		2 treatments per day	7 treatments per day	24 treatments per day
1	POAA treated, Log Reduction <sup>b</sup>	0	0	0.2
2	POAA-POOA treated, Log Reduction <sup>b</sup>	<0.1	2.2	1.9
3	POAA treated, % Germination <sup>c</sup>	>95%	>95%	< 50 %
4	POAA-POOA treated, % Germination <sup>c</sup>	>95%	>95%	< 80 %

- a) The water control was used as the background basis for the treatment efficiency Log reduction at the end of the treatment cycle.
- b) Log reduction vs. the water control.
- c) % Germination as per visual evaluation of % non-sprouted seeds.

5

#### **Working Example 9:**

##### **Hydroponic POAA and POAA-POOA Treatments of Alfalfa Sprouts With a Pre-dosing Misting Procedure**

The objective of this example was to evaluate microbial reduction using a 10 predosing of a peracid(s), followed by subsequent water-only mistings during the daily Hydroponic growth of alfalfa sprouts; against natural bacterial flora. Again, this allows for a lower overall dosage profile of peracids.

Using the above microbial technique, the alfalfa sprouts were soaked in 80 15 ppm of POAA or POAA-POOA solutions for 16 hours. One sample was soaked in water as a control. The following morning the alfalfa sprouts were placed into a 20 sterile petri dish by evenly spreading the seeds on the bottom of the dish. The petri dishes were covered with cheese cloth for the growing procedure.

During growing (days 1-4) the alfalfa seeds were treated from 8:00 AM to 3:00 PM on an hourly basis by misting with water only (no peracids in these 20 subsequent Hydroponic growth cycles). Likewise, the water control was misted with water. Microbial samplings were taken at 4:00 PM each of the 4 treatment days.

The results of Table 9 demonstrate that the peracid systems impart a residual antimicrobial effect over the entire Hydroponic growing cycle. Surprisingly, the 25 mixed peracid system yields a much improved efficacy versus the peracetic-alone formula; i.e., while peracetic acid can impart an initial (day 1) microbial reduction ( $> 1-\log$ ) vs. the control (cf., experiment 1,2), it fails after the 2nd growing day. Conversely the mixed peracetic-perooctanoic acid (POAA-POOA) system yields 30 continuous microbial control ( $> 2-\log$ ) over the entire sprout growing time (experiment 3) even though the growing seeds were only inoculated in the peracid formula. For all the experiments the germination rate of the seed was greater than 95%.

Table 9  
 Peracetic Acid (POAA) and Peracetic-Peroctanoic (POAA-POOA) Treatments of Alfalfa Sprouts Using A Single Mistning Procedure

Aerobic Plate Count Results

Sample Identification	CFU/ml	Day 1		Day 2		Day 3		Day 4	
		CFU/ml	Log R <sup>a</sup>	CFU/ml	Log R <sup>a</sup>	CFU/ml	Log R <sup>a</sup>	CFU/ml	Log R <sup>a</sup>
1 Water Control	2.0 x 10 <sup>6</sup>	0 <sup>b</sup>		7.0 x 10 <sup>6</sup>	0 <sup>b</sup>	5.8 x 10 <sup>6</sup>	0 <sup>b</sup>	5.2 x 10 <sup>6</sup>	0 <sup>b</sup>
2 peracetic acid (POAA)	9.4 x 10 <sup>3</sup>	1.3		1.8 x 10 <sup>6</sup>	0.6	2.9 x 10 <sup>6</sup>	0.3	3.9 x 10 <sup>6</sup>	0.1
3 peracetic-peroctanoic acids (POAA-POOA)	1.8 x 10 <sup>3</sup>	2.1		8.0 x 10 <sup>3</sup>	2.9	1.0 x 10 <sup>6</sup>	2.8	5.0 x 10 <sup>5</sup>	2.0
	80 ppm								

a) Log R = log reduction.  
 b) The water control was used as the background basis for the treatment efficiency Log reductions.

NOT TO BE TAKEN INTO ACCOUNT FOR THE PURPOSE OF INTERNATIONAL  
PROCESSING

14. The process of claim 13, wherein the hydrotrope is n-octanesulfonate, a xylene sulfonate, a naphthalene sulfonate, an amine oxide or a mixture thereof.

15. The process of claim 10, wherein the concentrate further comprises a chelating agent.

16. The process of claim 15, wherein the chelating agent is 1-hydroxyethylidene-1,1-diphosphonic acid.

17. A process for controlling fungi and microbial plant pathogens in growing plants comprising:

10 (a) diluting in an aqueous liquid a concentrate comprising:  
(i) about 1 to 20 wt-% of a C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid; and  
(ii) about 0.1 to 20 wt-% of an aliphatic C<sub>8</sub>-C<sub>12</sub> peroxy carboxylic acid;  
15 (iii) about 5 to 40 wt-% of a C<sub>2</sub>-C<sub>4</sub> carboxylic acid;  
(iv) about 1 to 20 wt-% of an aliphatic C<sub>8</sub>-C<sub>12</sub> carboxylic acid; and  
(v) about 1 to 30 wt-% of hydrogen peroxide to form a solution; and  
20 (b) contacting said growing plants with said solution.

18. The process of claim 17, wherein the C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid is peroxyacetic acid.

19. The process of claim 17, wherein the C<sub>8</sub>-C<sub>12</sub> aliphatic peroxy carboxylic acid is peroxyoctanoic acid.

25 20. The process of claim 17, wherein the concentrate further comprises about 1 to 15 wt-% of a hydrotrope.

21. The process of claim 17, wherein the hydrotrope is n-octanesulfonate, a xylene sulfonate, a naphthalene sulfonate, an amine oxide or a mixture thereof.

30 22. The process of claim 17, wherein the concentrate further comprises a chelating agent.

23. The process of claim 22, wherein the chelating agent is 1-hydroxyethylidene-1,1-diphosphonic acid.

35 24. The process of claim 17, wherein said C<sub>2</sub>-C<sub>4</sub> carboxylic acid is acetic acid.

25. The process of claim 17, wherein said aliphatic C<sub>8</sub>–C<sub>12</sub> carboxylic acid is octanoic acid.

26. A method of growing at least one plant on a hydroponic substrate in a hydroponic liquid supply medium to produce usable fruit or vegetable products with 5 reduced microbial contamination, the method comprising:

(a) establishing growing and living plant tissue in the hydroponic substrate;

10 (b) contacting the living plant tissue, the hydroponic substrate and the hydroponic liquid with a dilute aqueous solution comprising an effective amount of a C<sub>2</sub>–C<sub>12</sub> percarboxylic acid; and

(c) harvesting an improved product.

27. The method of claim 26 wherein the percarboxylic acid comprises peracetic acid.

28. The method of claim 26 wherein the percarboxylic acid comprises a 15 mixture of a C<sub>2</sub>–C<sub>4</sub> and a C<sub>5</sub>–C<sub>12</sub> aliphatic percarboxylic acid.

29. The method of claim 26 wherein the aqueous solution comprises about 4 to 100 parts per million of a C<sub>2</sub>–C<sub>4</sub> percarboxylic acid and about 1 to about 100 parts per million of an aliphatic C<sub>5</sub>–C<sub>12</sub> percarboxylic acid.

30. The method of claim 26 wherein the percarboxylic acid comprises a 20 mixture of peroxyacetic acid and peroxyoctanoic acid.

31. The process of claim 26 wherein the living tissue comprises a germinating seed.

32. The method of claim 26 wherein the living tissue comprises a growing tuber.

25 33. The method of claim 26 wherein the plant tissue comprises a growing dicotyledon.

34. The method of claim 26 wherein the plant tissue comprises a growing monocotyledonis plant.

35. The method of claim 26 wherein the living tissue comprises a plant 30 cutting.

36. The method of claim 26 wherein the plant tissue comprises rooting stock and a graft.

## AMENDED CLAIMS

[received by the International Bureau on 9 September 1999 (09.09.99);  
claims 1-13 added; remaining claims unchanged (1 page)]

1. A method of controlling microbial pathogens on living plant tissue comprising treating said plant tissue with a dilute aqueous solution comprising an effective amount of a C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid and an aliphatic C<sub>8</sub>-C<sub>12</sub> peroxy carboxylic acid.
- 5 2. The method of claim 1 wherein the plant tissue comprises a seed.
3. The method of claim 1 wherein the plant tissue comprises a tuber.
4. The method of claim 1 wherein the plant tissue comprises a growing plant.
- 10 5. The method of claim 1 wherein the plant tissue comprises a cutting.
6. The method of claim 1 wherein the plant tissue comprises root stock.
7. The method of claim 1, wherein the aqueous solution comprises:
  - (a) at least about 4 parts per million (ppm) of a C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid; and
  - (b) at least about 1 parts per million (ppm) of an aliphatic C<sub>8</sub>-C<sub>12</sub> peroxy carboxylic acid.
- 15 8. The method of claim 1, wherein said C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid is peroxyacetic acid.
9. The method of claim 1 wherein said C<sub>8</sub>-C<sub>12</sub> aliphatic peroxy carboxylic acid is peroxyoctanoic acid.
- 20 10. A process for controlling fungi and microbial plant pathogens in growing plants comprising:
  - (a) diluting in an aqueous liquid a concentrate comprising:
    - (i) about 1 to 20 wt-% of a C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid;
  - 25 and
  - (ii) about 0.1 to 20 wt-% of an aliphatic C<sub>8</sub>-C<sub>12</sub> peroxy carboxylic acid to form a solution; and
  - (b) contacting said growing plants with said solution.
11. The process of claim 10, wherein the C<sub>2</sub>-C<sub>4</sub> peroxy carboxylic acid is peroxyacetic acid.
- 30 12. The process of claim 10, wherein the C<sub>8</sub>-C<sub>12</sub> aliphatic peroxy carboxylic acid is peroxyoctanoic acid.
13. The process of claim 10, wherein the concentrate further comprises about 1 to 15 wt-% of a hydrotrope.

# INTERNATIONAL SEARCH REPORT

Inter	nal Application No
PCT/US 99/06010	

**A. CLASSIFICATION OF SUBJECT MATTER**  
 IPC 6 A01N37/16 A01G31/00 // (A01N37/16, 37:16, 59:00)

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A01N

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	<p>WO 93 01716 A (ECOLAB INC) 4 February 1993            see page 1 - page 3, line 2            see page 4, line 24 - page 9, line 9            see page 9, line 29 - page 10, line 4            see claims 1,11,12,15,18,52,53,61;            examples 1-30            &amp; US 5 200 189 A            cited in the application            ---</p>	17-36
Y	<p>EP 0 242 990 A (ALBRIGHT &amp; WILSON)            28 October 1987            cited in the application            see page 1            see page 3, line 20 - line 39; example 1            ---            -/---</p>	17-36

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

\* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- "&" document member of the same patent family

Date of the actual completion of the international search

30 June 1999

Date of mailing of the international search report

12/07/1999

Name and mailing address of the ISA  
 European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
 Fax: (+31-70) 340-3016

Authorized officer

Muellers, W

1

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	EP 0 361 955 A (ALBRIGHT & WILSON) 4 April 1990 see page 2, line 1 - line 43; examples 1,3 & US 5 168 655 A cited in the application ---	26-36
A	WO 94 21122 A (ECOLAB INC) 29 September 1994 see page 1, paragraph 1 see page 6, line 11 - page 10, line 19 see page 12, line 11 - line 26 see working examples 1 and 7 see claims 1-5 ---	17-36
A	WO 94 23575 A (ECOLAB INC) 27 October 1994 see page 1, line 5 - line 11 see page 2, line 26 - page 11, line 23; claims 1,5,6,11,12,15; examples 1-32 -----	17-36

## INTERNATIONAL SEARCH REPORT

Int. application No.  
PCT/US 99/06010

### Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1.  Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:
  
2.  Claims Nos.: 1-16 because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:  
see FURTHER INFORMATION sheet PCT/ISA/210
  
3.  Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

### Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1.  As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
  
2.  As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
  
3.  As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
  
4.  No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

#### Remark on Protest

The additional search fees were accompanied by the applicant's protest.

No protest accompanied the payment of additional search fees.

**FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210**

Claims 1-13 were omitted from the application when filed. The International Searching Authority refused to authorize a rectification request under Rule 91.1 PCT thereby referring to paragraph (c) of this Rule.

Claims 14-16 depend directly or indirectly on omitted claims 13 and 10 respectively.

**FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210**

Claims Nos.: 1 - 16

Claims 1-13 were omitted from the application when filed. The International Searching Authority refused to authorize a rectification request under Rule 91.1 PCT thereby referring to paragraph (c) of this Rule.

Claims 14-16 depend directly or indirectly on omitted claims 13 and 10 respectively

**INTERNATIONAL SEARCH REPORT**

Information on patent family members

Interr. Application No

PCT/US 99/06010

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
WO 9301716	A 04-02-1993	US 5200189 A		06-04-1993
		AT 161142 T		15-01-1998
		AU 652274 B		18-08-1994
		AU 2176992 A		23-02-1993
		CA 2108177 A		24-01-1993
		CN 1068705 A		10-02-1993
		DE 69223628 D		29-01-1998
		DE 69223628 T		09-04-1998
		DK 597877 T		09-02-1998
		EP 0597877 A		25-05-1994
		ES 2112908 T		16-04-1998
		FI 940231 A		17-03-1994
		JP 2874041 B		24-03-1999
		JP 6510526 T		24-11-1994
		MX 9202599 A		01-01-1993
		NO 934217 A		22-11-1993
		US 5718910 A		17-02-1998
		US 5437868 A		01-08-1995
		US 5489434 A		06-02-1996
		US 5314687 A		24-05-1994
EP 0242990	A 28-10-1987	AU 601404 B		13-09-1990
		AU 7044087 A		24-09-1987
		GB 2187958 A, B		23-09-1987
EP 0361955	A 04-04-1990	AT 118970 T		15-03-1995
		AU 624420 B		11-06-1992
		AU 4179689 A		05-04-1990
		CA 1336883 A		05-09-1995
		DE 68921399 D		06-04-1995
		DE 68921399 T		21-09-1995
		DK 479389 A		30-03-1990
		ES 2072907 T		01-08-1995
		GB 2224441 A, B		09-05-1990
		US 5168655 A		08-12-1992
WO 9421122	A 29-09-1994	US 5409713 A		25-04-1995
		AU 675975 B		27-02-1997
		AU 5088893 A		11-10-1994
		CA 2156299 A		29-09-1994
		CN 1092385 A		21-09-1994
		NZ 255871 A		27-08-1996
		US 5674538 A		07-10-1997
		US 5683724 A		04-11-1997
WO 9423575	A 27-10-1994	US 5437868 A		01-08-1995
		AT 167981 T		15-07-1998
		AU 676902 B		27-03-1997
		AU 6586794 A		08-11-1994
		DE 69411554 D		13-08-1998
		DE 69411554 T		24-12-1998
		EP 0693876 A		31-01-1996
		NZ 265418 A		26-07-1996
		US 5489434 A		06-02-1996